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Key indicators

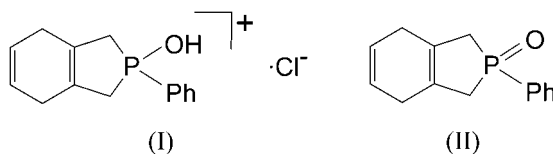
Single-crystal X-ray study
 $T = 198$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.025
 wR factor = 0.066
Data-to-parameter ratio = 10.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2-Hydroxy-2-phenyl-4,7-dihydroisophosphindolinium
chloride

The title compound, $\text{C}_{14}\text{H}_{16}\text{OP}^+\cdot\text{Cl}^-$ (systematic name: 2-hydroxy-2-phenyl-1,3,4,7-tetrahydroisophosphindinium chloride), displays a nearly planar isophosphindolinium skeleton, where a significant deviation from planarity is observed only for the P atom. Strong $\text{O}-\text{H}\cdots\text{Cl}$ bonds result in ion pairs, but an extended hydrogen-bonding network is not present.

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Comment

The title compound, (I), is an intermediate in one of the avenues pursued in the synthesis and characterization of tricarbonyl(η^5 -isophosphindolyl)manganese, the first complex bearing the isophosphindolyl skeleton (Decken *et al.*, 2004).



The isophosphindolinium ring system in (I) is not completely planar; the eight C atoms of the bicyclic unit fit a least-squares plane to within 0.037 (2) Å, but the P atom deviates by 0.378 (2) Å from this plane. This results in a folding angle of 17.6 (1)° with the C1/P/C3 plane. However, the distortion from planarity is not as pronounced as in the closely related compound 2-oxo-2-phenyl-1,3,4,7-tetrahydroisophosphindole, (II), (Decken *et al.*, 2004), where the deviation of the P atom is 0.484 (2) Å, leading to a folding angle of 22.58 (6)°. In addition, the envelope is folded toward the hydroxy group and away from the phenyl group in (I), opposite to the conformation adopted in (II). As a result, torsion angles are 99.6 (1) (C4a-C3-P-O), 102.8 (1) (C7a-C1-P-O), 145.7 (1) (C4a-C3-P-C8) and 145.2 (1)° (C7a-C1-P-8) in compound (I), while the corresponding angles are 135.3 (1), 134.6 (1), 88.9 (1) and 87.7 (1)° in (II). The title compound is ionic, with Cl^- anions located between sheets of the organic cations and held by strong $\text{O}-\text{H}\cdots\text{Cl}$ hydrogen bonds [1.87 (3) Å]. However, a much shorter hydrogen bond is observed in $\text{Ph}_3\text{P}-\text{O}-\text{H}\cdots\text{Cl}$, (III) [1.63 (4) Å; Haupt *et al.*, 1977], the only hydroxyphosphonium chloride previously characterized by X-ray crystallography. The P-O bond length [1.5516 (14) Å] is comparable to the corresponding distances in (III) [1.517 (2) Å], $\text{Ph}_3\text{P}-\text{O}-\text{H}\cdots\text{Br}$ [1.550 (6) Å; Lane *et al.*, 1992] and ${}^i\text{Pr}_3\text{P}-\text{O}-\text{H}\cdots\text{I}$ [1.573 (2) Å; Ruthe *et al.*, 2000]. The P atom in (I) is in a tetrahedral environment and the phenyl group is almost orthogonal to the isophosphindolinium fragment [87.56 (4)°], similar to the geometry observed in (II) [87.06 (6)°].

Experimental

The title compound is an intermediate in the published synthesis of 2-phenyl-1,3,4,7-tetrahydroisosphindole 2-oxide (Middlemas & Quin, 1979). ³¹P {¹H} NMR (121.356 MHz, CDCl₃): δ 62.0. ¹H NMR (399.945 MHz, CDCl₃): δ 7.75 (dd, *J* = 7.2, 7.2 Hz, 2H), 7.58 (d, *J* = 7.2 Hz, 1H), 7.51 (dd, *J* = 7.2, 7.2 Hz, 2H), 5.80 (s, 2H), 2.98 (t, *J* = 16.8 Hz, 2H), 2.75 (m, 6H).

Crystal data

C₁₄H₁₆OP⁺·Cl⁻

M_r = 266.69

Orthorhombic, *Pna*2₁

a = 14.9111 (15) Å

b = 8.7445 (9) Å

c = 10.4465 (10) Å

V = 1362.1 (2) Å³

Z = 4

D_x = 1.300 Mg m⁻³

Mo *K*α radiation

Cell parameters from 4853

reflections

θ = 2.7–25.5°

μ = 0.38 mm⁻¹

T = 198 (2) K

Irregular fragment, colourless

0.50 × 0.28 × 0.25 mm

Data collection

Bruker AXS SMART 1000/P4
diffractometer

ω and *φ* scans

Absorption correction: multi-scan
(*SADABS*; Sheldrick, 1996)

T_{min} = 0.816, *T_{max}* = 0.910

6519 measured reflections

2254 independent reflections

2180 reflections with *I* > 2σ(*I*)

R_{int} = 0.020

θ_{max} = 25.0°

h = -17 → 16

k = -9 → 10

l = -12 → 12

Refinement

Refinement on *F*²

R [*F*² > 2σ(*F*²)] = 0.025

wR (*F*²) = 0.066

S = 1.06

2254 reflections

218 parameters

All H-atom parameters refined

w = 1/[σ²(*F_o*²) + (0.0498*P*)²
+ 0.02*P*]

where *P* = (*F_o*² + 2*F_c*²)/3

(Δσ)_{max} = 0.001

Δρ_{max} = 0.29 e Å⁻³

Δρ_{min} = -0.11 e Å⁻³

Absolute structure: Flack (1983),

997 Friedel pairs

Flack parameter = 0.04 (6)

Table 1

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O—H14...Cl ⁱ	0.98 (3)	1.87 (3)	2.8449 (16)	172 (2)

Symmetry code: (i) $\frac{3}{2} - x, y - \frac{1}{2}, \frac{1}{2} + z$.

H atoms were located in Fourier difference maps and refined with isotropic displacement parameters; C—H distances are in the range 0.90 (2)–0.99 (2) Å.

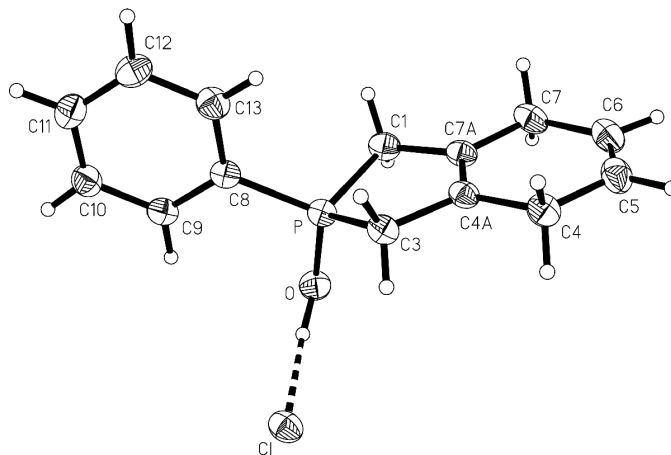


Figure 1

A view of the title compound, with displacement ellipsoids drawn at the 30% probability level. The dashed line indicates a hydrogen bond.

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

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